

New autonomous sensors for underway measurements

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April 2013, Helsinki

5th FerryBox Workshop

Motivation

Topics of interest

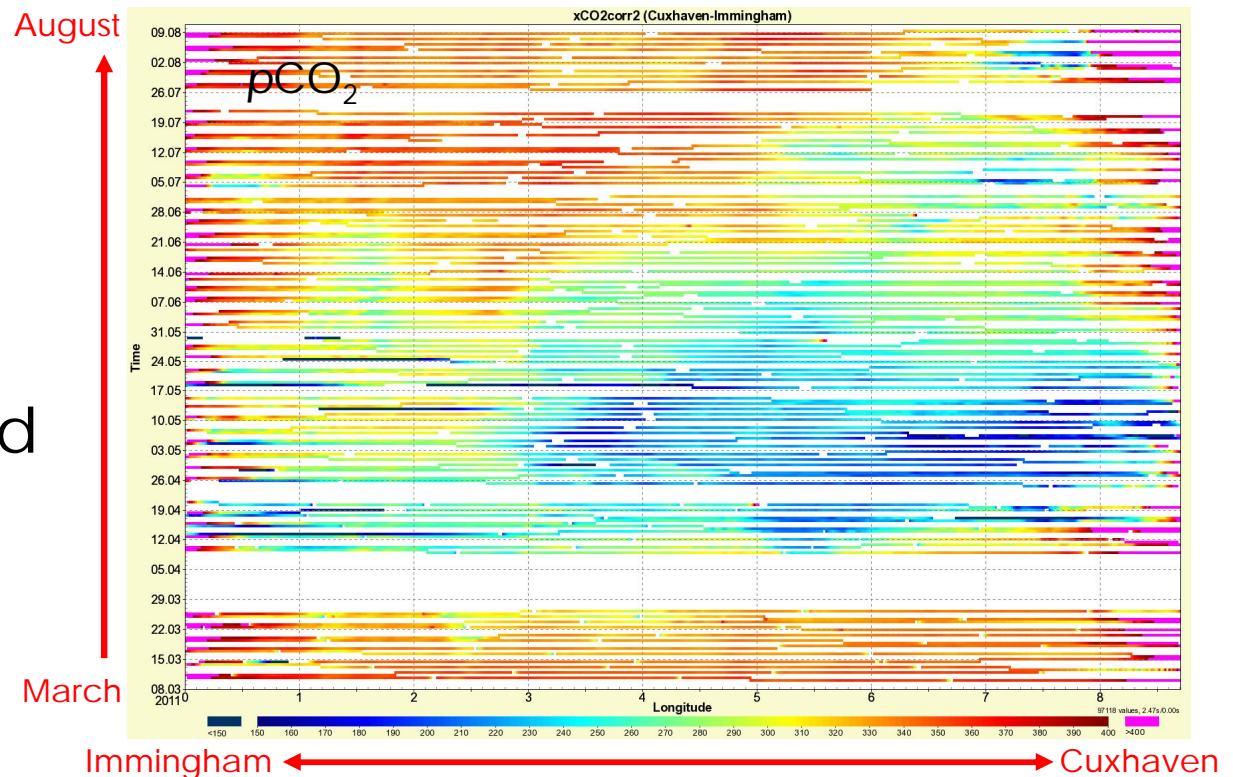
- Marine Strategy Framework Directive (MSDF)
 - Eutrophication Nutrients
 - Harmful algae blooms (HABs) Identification of algae

 - Primary production O_2 , pCO_2 , Chl-a

 - Ocean acidification
 - Alkalinity transport pH
 - Sinks / sources for CO_2 Total alkalinity (TA)
 - Feedbacks to the rising pCO_2 , pH
 - atmospheric CO_2 concentration pCO_2 , pH, TA
 - concentration
- From physical parameters to biogeochemistry

FerryBoxes as platforms

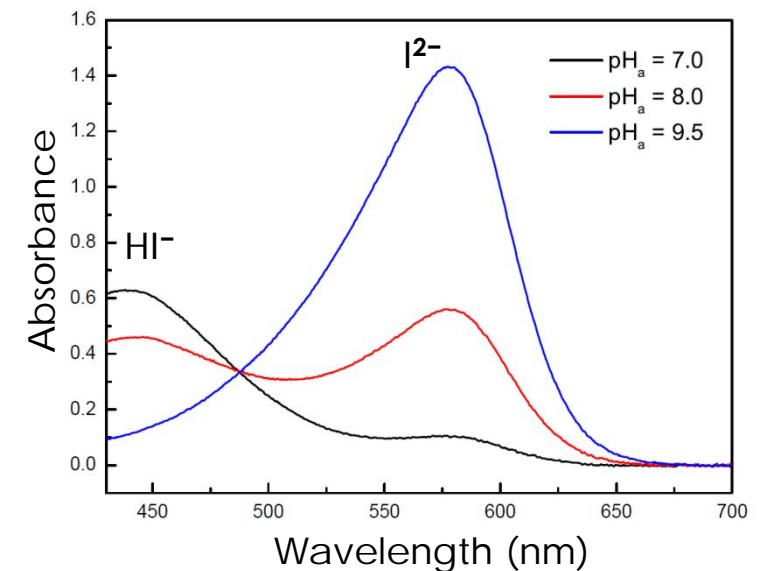
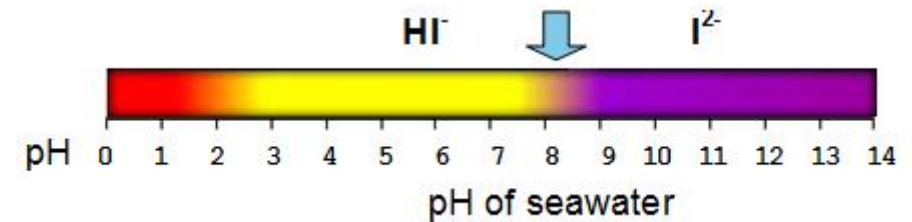
- cost effective and good to handle on SOO
- reduced demands on autonomy for new sensors developments
- high spatial and temporal resolution
- long-term records and seasonally resolution
- tracking of short-term biological processes



Sensors

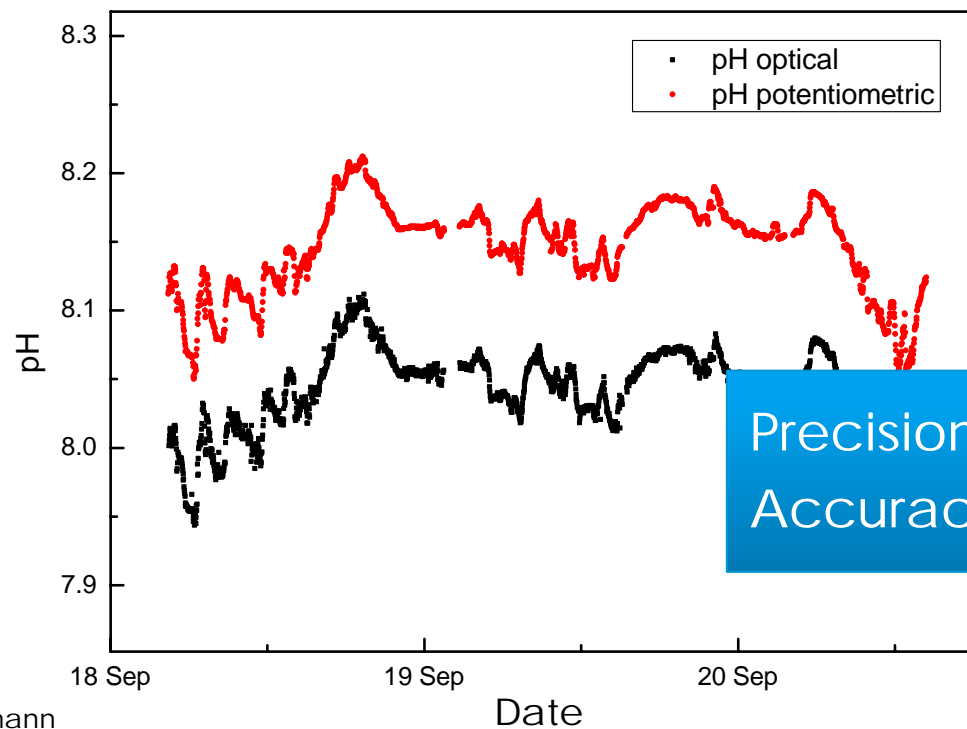
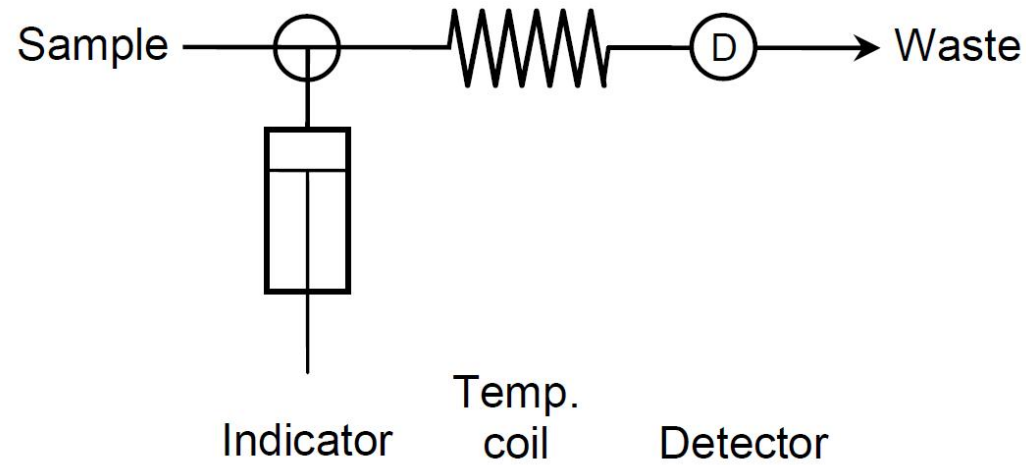
pH – Principle

- FIA system using an indicator dye *m*-Cresol purple
- Determination of the concentration of the indicator acid (HI⁻) / base (I²⁻) due to different absorption spectra
- Calculation of the pH value using Henderson–Hasselbalch equation

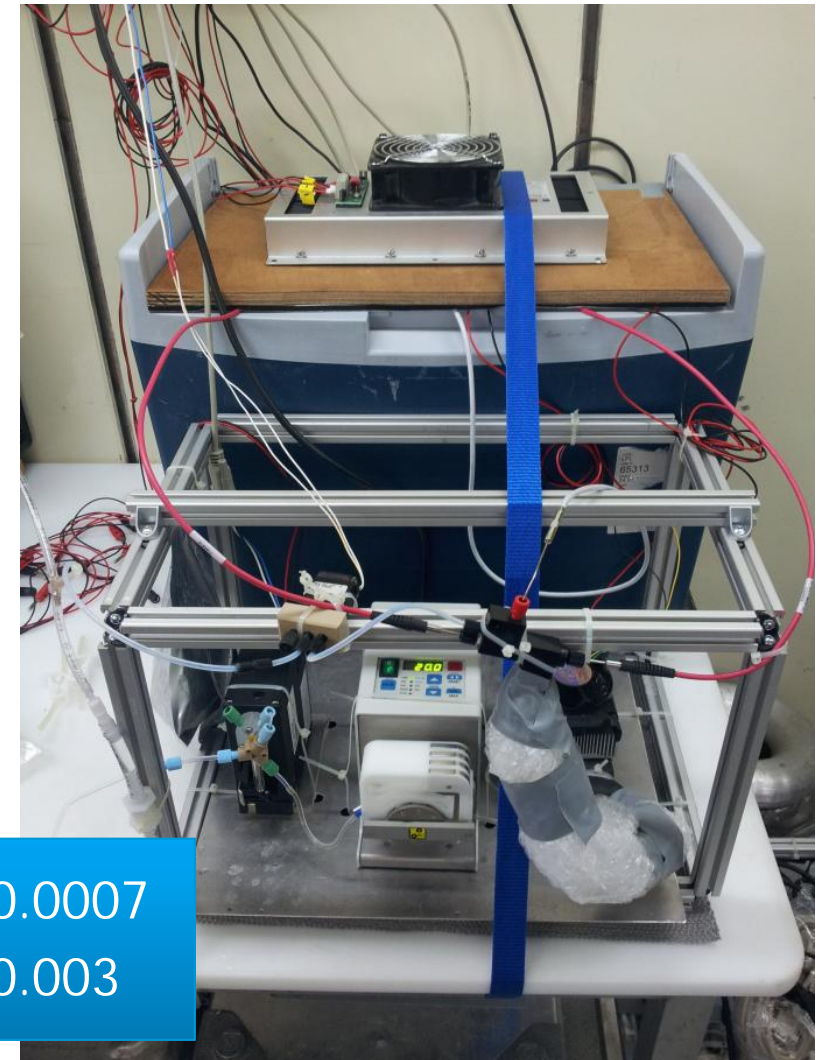


$$\text{pH} = \text{p}K_a + \log_{10} \frac{[\text{I}^{2-}]}{[\text{HI}^{-}]}$$

pH – Setup

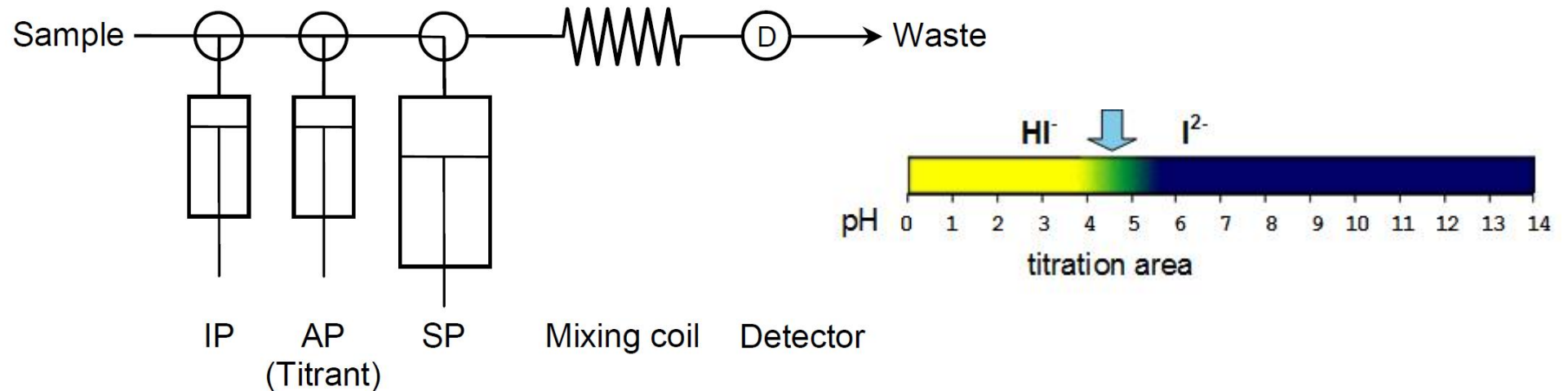


Precision ± 0.0007
 Accuracy ± 0.003



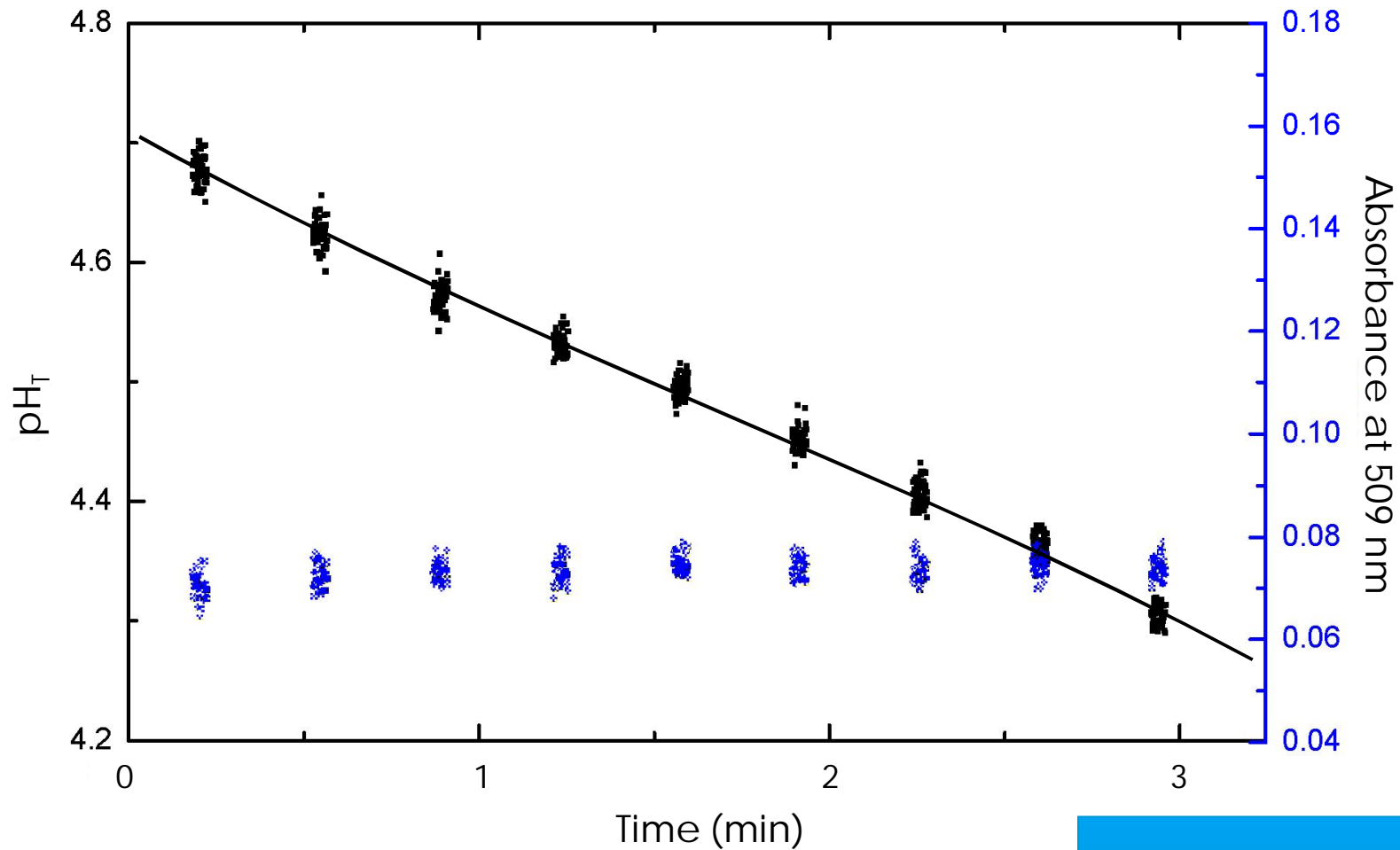
Aßmann et al., 2011

Total alkalinity (TA)

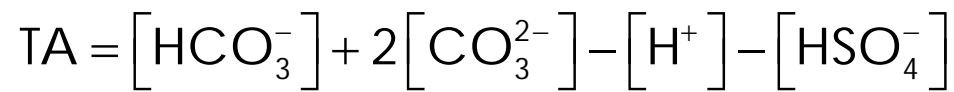


- Closed-cell titration with HCl (syringe aspirates simultaneously sample water, indicator dye and HCl)
- Monitoring of the titration curve (pH range of 3.5 to 5.5) by the acid-base indicator dye (Bromocresol green)
- Calculation of TA by a least-squares procedure based on a non-linear curve fitting approach

Total alkalinity (TA)

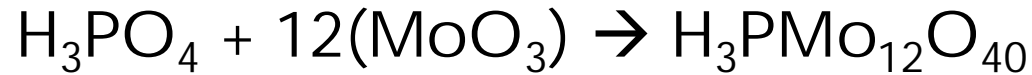


Precision $\pm 5 \mu\text{mol/kg}$
 Accuracy $\pm 1 \mu\text{mol/kg}$

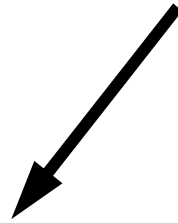


$$0 = \boxed{\text{TA} - \text{DIC}} \left(\frac{K_1[\text{H}^+]_T + 2K_1K_2}{[\text{H}^+]_T^2 + K_1[\text{H}^+]_T + K_1K_2} \right) + S_T \left(\frac{1}{1 + K_S/[\text{H}^+]_F} \right) + \left(\frac{M_{\text{SW}} + M_A}{M_{\text{SW}}} \right) \left(\frac{[\text{H}^+]_T}{Z} - \frac{K_W}{[\text{H}^+]_T} \right) - \frac{M_A}{M_{\text{SW}}} N_A$$

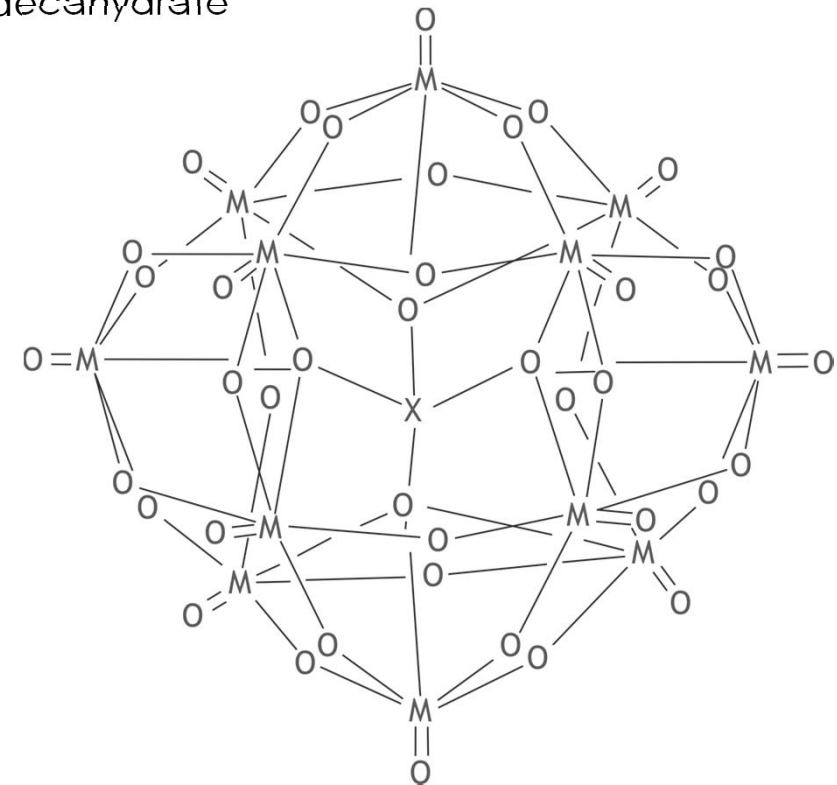
Phosphate – Principle



Phosphate + Molybdate \rightarrow 12-Molybdophosphate hexadecahydrate

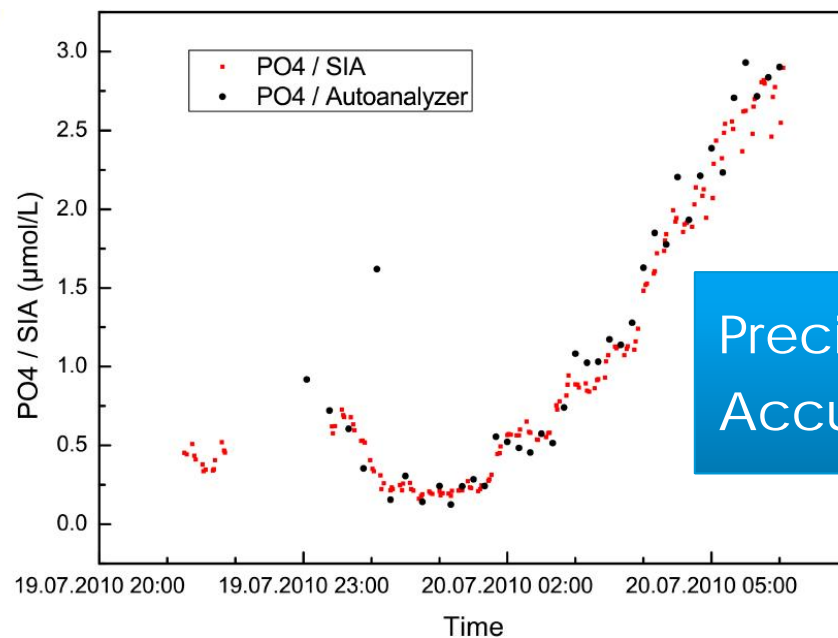
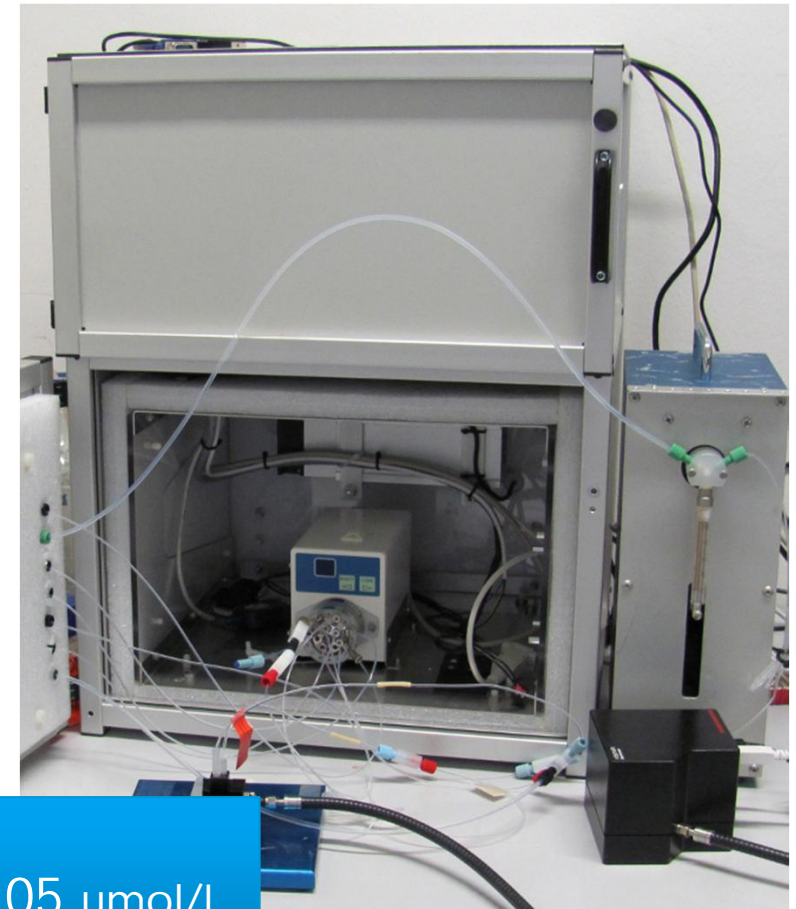
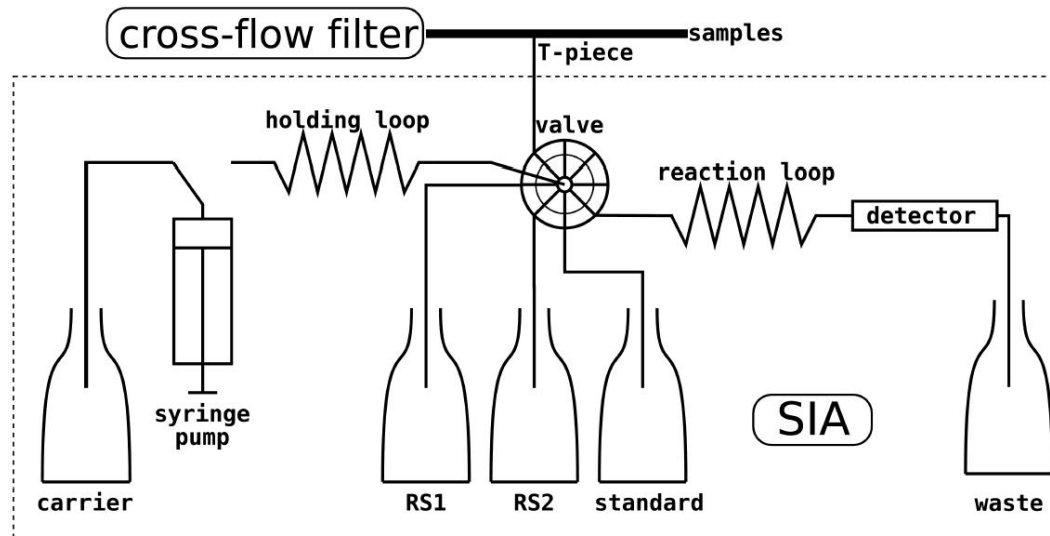


- Fluorescence quenching of Rhodamine 6G
- Excitation at 470 nm
Emission at 550 nm



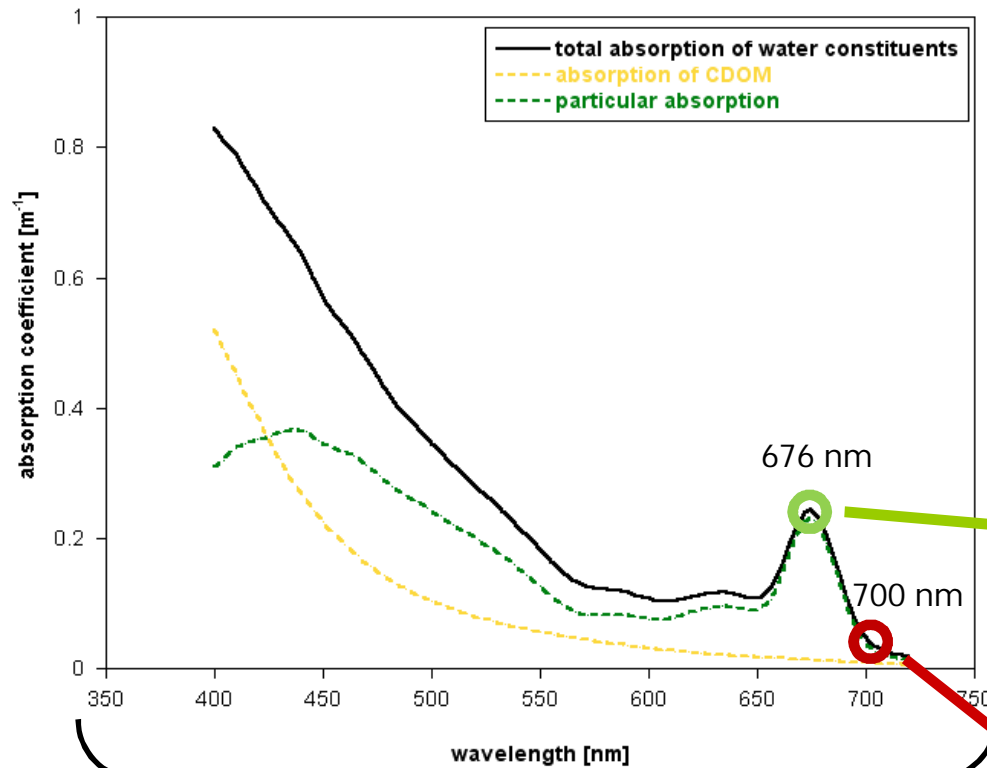
Keggin structure

Phosphate – Setup / Data



Frank & Schroeder, 2007
Frank et al., 2006

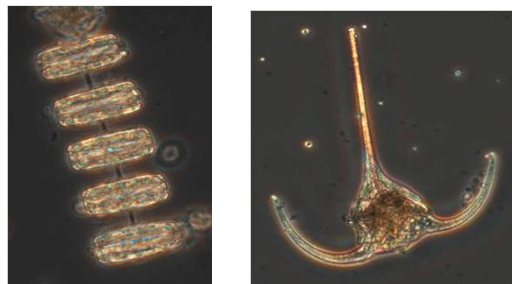
ft-PSICAM – Principle



flow through –
Point Source Integrating
Cavity Absorption Meter

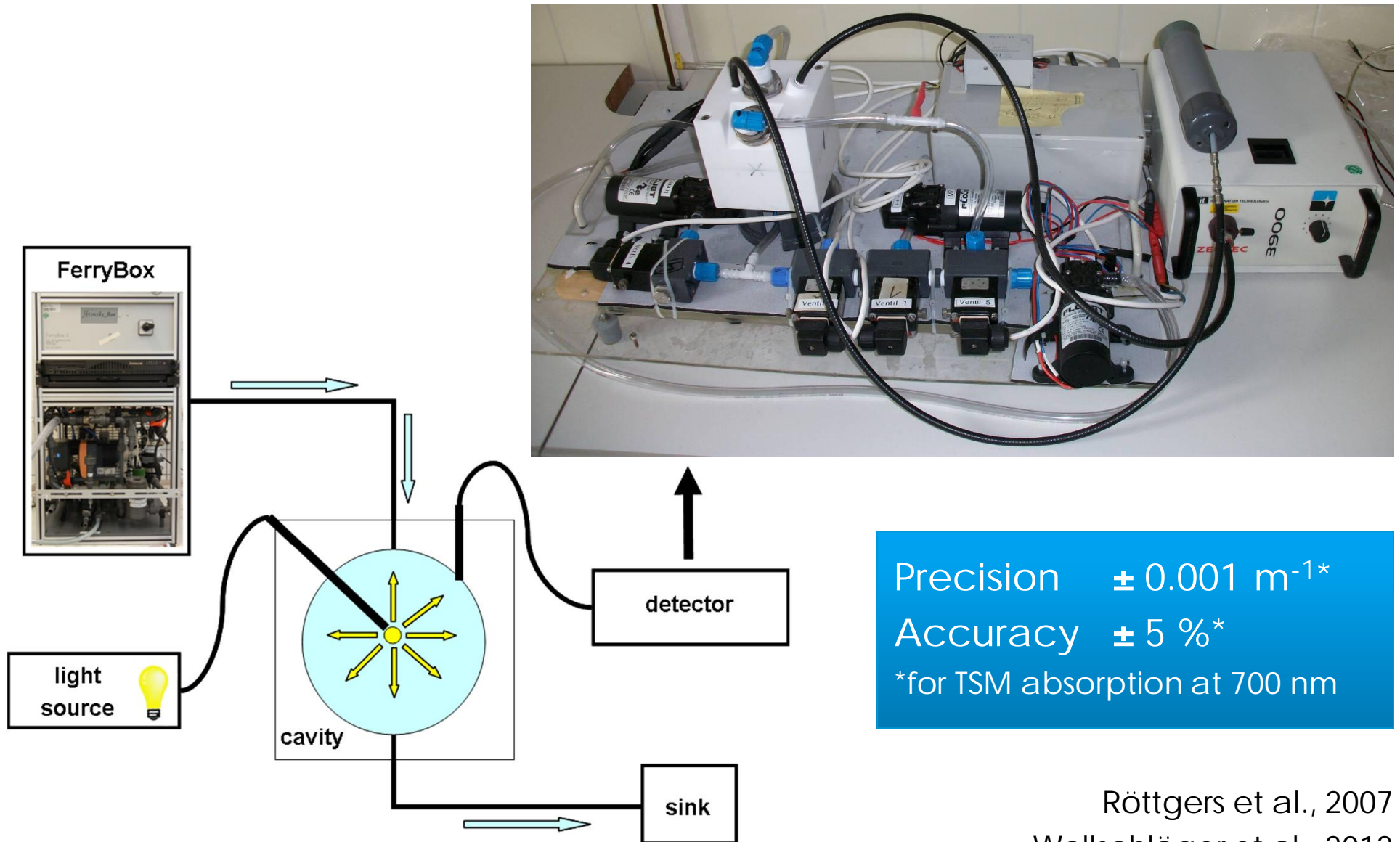
Absorption of
chlorophyll-a

Total suspended
matter (TSM)



Identification of algae groups

PSICAM – Setup

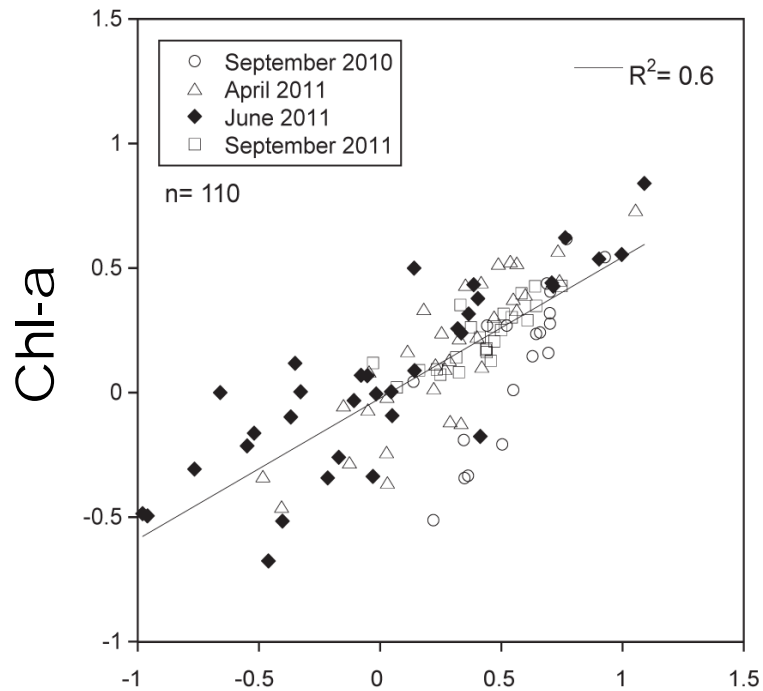


Precision $\pm 0.001 \text{ m}^{-1}$ *
Accuracy $\pm 5 \%$ *
*for TSM absorption at 700 nm

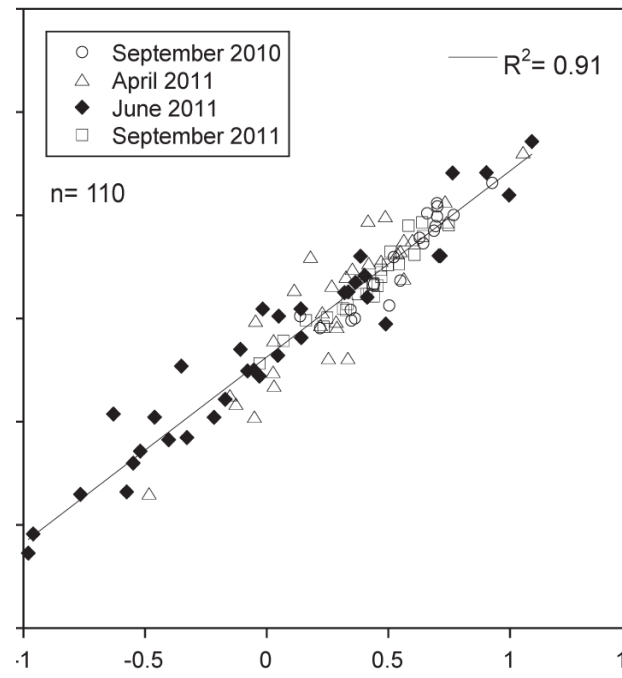
Röttgers et al., 2007
Wollschläger et al., 2012

PSICAM – Data

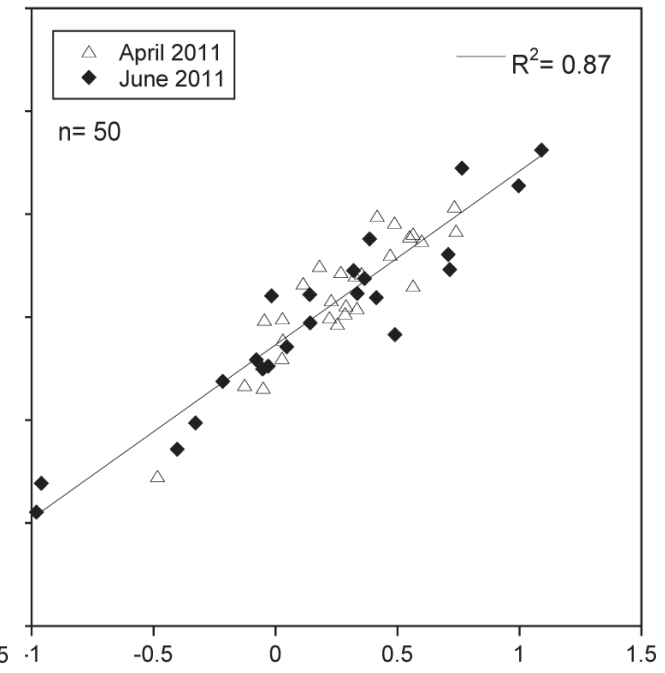
Chl-a Fluorescence



Chl-a PSICAM



Chl-a ft-PSICAM



Chl-a from HPLC

Summary

	pH	TA	PO ₄ ³⁻	Absorption (PSICAM, 700 nm)
Accuracy	± 0.003 (with CRM)	± 1 µmol/kg	± 0.05 µmol/kg	± 5 %
Precision	± 0.0007	± 5 µmol/kg		± 0.001 m ⁻¹
Meas. Cycle	1 min	5 min	1 min	1 min
Range	7.5–9.0	1.8–2.5 mmol/kg	0.1–10 µmol/kg	0.001–10 m ⁻¹
Method	absolute	absolute	relative	absolute

Summary

FerryBoxes are suitable platforms for sensor development and their final application.

Biochemical sensors are needed for monitoring and understanding the changes of the oceans.

- Nutrients for evaluation of the trophic status
- CO₂ parameters for assessment of ocean acidification
- Absorption measurements for tracking and identification of algae

Outlook

- Fully characterized carbonate system
- Alkalinity transport between sediment and seawater
into the ocean
- CO₂ air-sea fluxes (sinks / sources for CO₂)
- Quantification of production rates
- Comparison of productivity with estimates derived from
different parameters (CO₂, O₂, Chl-a, winter nutrient
stocks ...)
- Phytoplankton dynamics (e.g. seasonality ...)
- More reliable chlorophyll-a and turbidity data
- Differentiation of major algal groups by optical absorption
signatures

New autonomous sensors for underway measurements

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Thank you

5th FerryBox Workshop

References

- Aßmann, S. (2012). *Entwicklung und Qualifizierung autonomer Messsysteme für den pH-Wert und die Gesamtalkalinität von Meerwasser*. Christian-Albrechts-Universität zu Kiel. Retrieved from http://eldiss.uni-kiel.de/macau/receive/dissertation_diss_00010645
- Aßmann, S., Frank, C., & Körtzinger, A. (2011). Spectrophotometric high-precision seawater pH determination for use in underway measuring systems. *Ocean Science*, 7(5), 597–607. doi:10.5194/os-7-597-2011
- Röttgers, R., & Doerffer, R. (2007). Measurements of optical absorption by chromophoric dissolved organic matter using a point-source integrating-cavity absorption meter. *Limnology and Oceanography: Methods*, 5, 126–135. doi:10.4319/lom.2007.5.126
- Frank, C., & Schroeder, F. (2007). Using Sequential Injection Analysis to Improve System and Data Reliability of Online Methods: Determination of Ammonium and Phosphate in Coastal Waters. *Journal of Automated Methods and Management in Chemistry*, 2007, 49535. doi:10.1155/2007/49535
- Frank, C., Schroeder, F., Ebinghaus, R., & Ruck, W. (2006). Using sequential injection analysis for fast determination of phosphate in coastal waters. *Talanta*, 70(3), 513–517. doi:10.1016/j.talanta.2005.12.055
- Wollschläger, J., Grunwald, M., Röttgers, R., Petersen, W. (2012). Flow-through-PSICAM: A new approach for determining water constituents absorption continuously. *Ocean Dynamics* (subm.)